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Ion mobility as additional Metrological Value – the Invaluable Benefit of another Dimension in Hybrid Mass Spectrometry

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²Sigma-Aldrich Productions, Merck KGaA, Buchs, Switzerland

³Waters AG, Baden Dättwil, Switzerland

"lon path"

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- A small review of Ion Mobility
 - History
 - Robustness, time, space, matrix and platform independence
- Examples of IMS-HRMS:
- Summary

Firstly: what on earth has the mutiny on the HMS Bounty (1789) to do with this talk????

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Ion Mobility Spectroscopy Standalone: Screening Instrument for Security/Military uses

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Sample flow

Repelling rings Ion shutter Drift gas flow

Sample

Desorber heater

https://www.sas.upenn.edu/~lindamf/spectIMS.html



https://www.bruker.com/en/products-and-solutions/cbrne-detectors/ims.html

https://www.asiatraveltips.com/news20/412-AirportSecurity.shtml

The IONSCAN 500DT, certified on the TSA Qualified Products List for security screening, utilizes Ion Mobility Spectroscopy (IMS) to perform trace analysis of explosives in seconds. Easy to use, the operators can detect a wide range of military, commercial and homemade threats.



https://www.masatech.eu/blog

GDA - Personal



Historical Developments in Ion Mobility (IM) Technologies

1890-

1900

1910

1920-

1930

1940-

1950-

1960

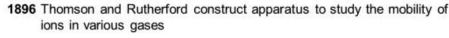
1970

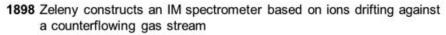
1980-

1990

2000

-2010-





- 1911 Millikan develops apparatus for measuring the size-to-charge ratio of
- 1928 Tyndall constructs a precision ion mobility drift tube spectrometer using a dual ion gate design
- 1930 Tyndall improves mobility measurements by using pure drift gases
- 1961 McDaniel couples ion mobility to a magnetic sector MS (IM-MS)
- 1963 McAfee and Edelson interface a drift tube orthogonally to a time-offlight mass spectrometer (IM-oTOF)
- 1964 Hasted and coworkers develop mass-selected ion mobility-mass spectrometry (MS-IM-MS)
- 1968 Dole develops ESI with ion mobility measurements
- 1970 first commercial ion mobility spectrometer (Plasma Chromatograph)
- 1975 first commercial DMA (Thermo-Systems)
- 1982 Lubman couples laser ionization with ion mobility
- 1982 Hill develops gas chromatography coupled to ion mobility
- 1989 Blanchard describes tandem IM strategies (IM/IM)
- 1990 introduction of FAIMS and DMS
- 1990 commercial portable IM spectrometers (several vendors)
- 1995 Bowers develops MALDI-IM-MS and variable-temperature IM
- 1996 Jarrold constructs a high resolution drift tube IM spectrometer
- 1998 Smith develops the electrodynamic ion funnel
- 2006 commercial traveling-wave IM-MS (Waters)
- 2011 trapped ion mobility coupled to MS developed (Bruker)
- 2014 commercial drift tube IM-MS (Agilent)



150 0 **Publications**

300

Ion Mobility

Ion Mobility-Mass Spectrometry

Temporally-Dispersive IM-MS

Spatially-Dispersive IM-MS

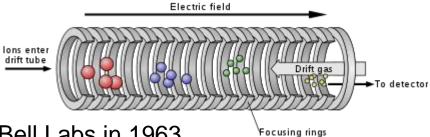
(Jody C. May and John A. McLean*; Anal Chem. 2015 Feb 3; 87(3): 1422-1436)

Ion Mobility Spectroscopy is OLDER than Mass Spectrometry!!

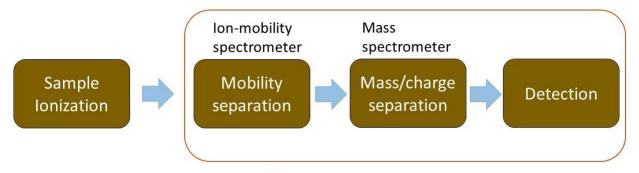


Discovered by JJ Thompson and Rutherford 1896, but developed by EW McDaniel in 1950-1960s. First drift tube IMS was designed in 1960s

McDaniel E, Martin DW, Barnes WS (1962). "Drift Tube-Mass Spectrometer for Studies of Low-Energy Ion-Molecule Reactions". *Review of Scientific Instruments*. **33** (1): 2–7. Bibcode:1962RScl...33....2M. doi:10.1063/1.1717656. ISSN 0034-6748.

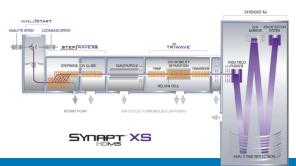


First Hybrid IMS TOF MS was designed by Bell Labs in 1963



Ion mobility spectrometry-mass spectrometry

First commercially available IMS QTOF 2006 (Waters)



What is Ion Mobility – Definition:

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terminal velocity *v* (of an ion) in a gasphase

$$V_o = KE$$

E is the electric field strength and K is known as the ion's mobility.

 $CCS = f\{K\}$, Linear Function of K

(CCS = Collisional Cross Section Size (in Area, Å²))

Ion-Mobility principle

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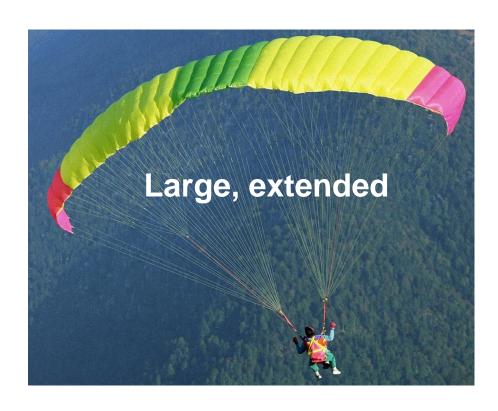
high mobility, more compact, less collisions with gas



low mobility, less compact, more collisions with gas

- CCS (~K) is an important distinguishing characteristic of an ion which is related to:
 - -chemical structure
 - -3-dimensional conformation
- CCS (~K) is a physicochemical property of an ion.

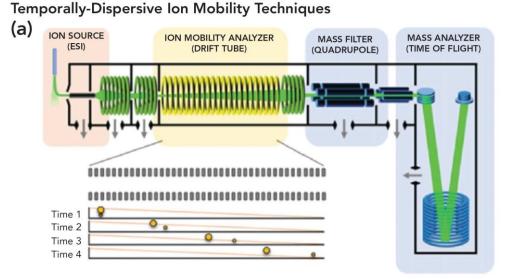


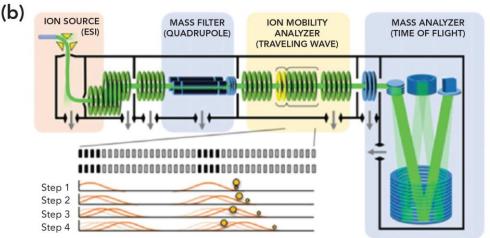


What is Collision Cross Section (CCS)?

How does IMS practically combine into a Mass Spectrometer:







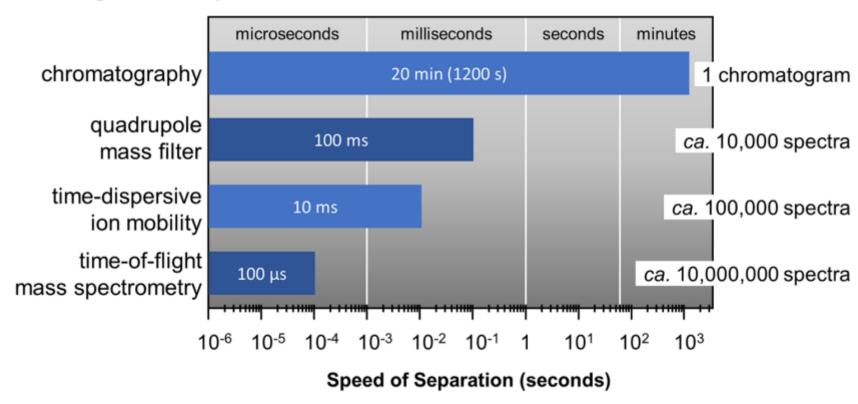
For Targeted methods or for prefiltering there are other techniques – like FAIMS or SelexionTM, but this is not in the scope of this talk

I.S. Krull, A.S. Rathore, Jared Auclair, LCGC North America, LCGC-11-01-2020, Volume 38, Issue 11, Pages: 619-624

The reason why IMS data collection fits in between Fast Chromatography (GC, SFC or LC) and (Q)TOF HRMS:

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Nesting of Analytical Timescales



(Jody C. May and John A. McLean*; Anal Chem. 2015 Feb 3; 87(3): 1422–1436)

3D resolution: Analysis strategy and sample complexity

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1: UPLC Peak 1-3 secs

1: Chromatographic Resolution

2: Ion Mobility Resolution

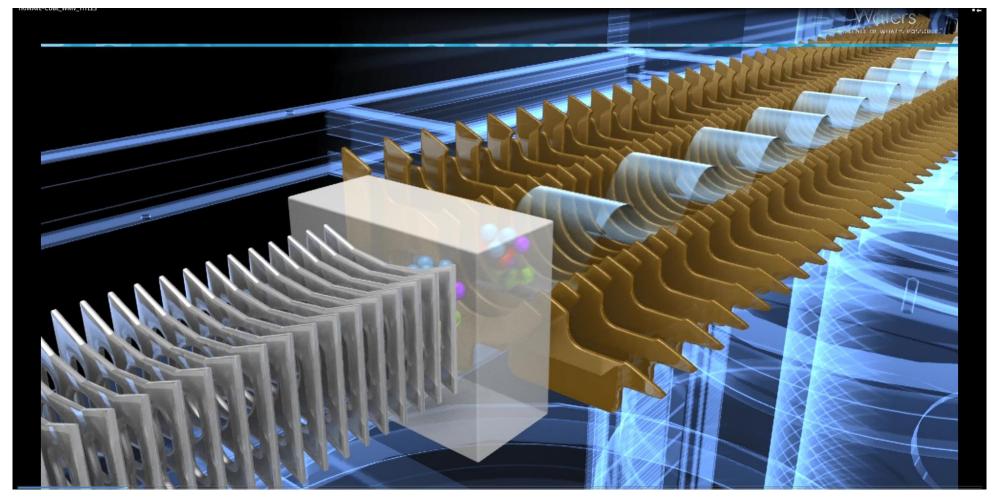
3: Mass Resolution



Scan range: m/z 50 -1200

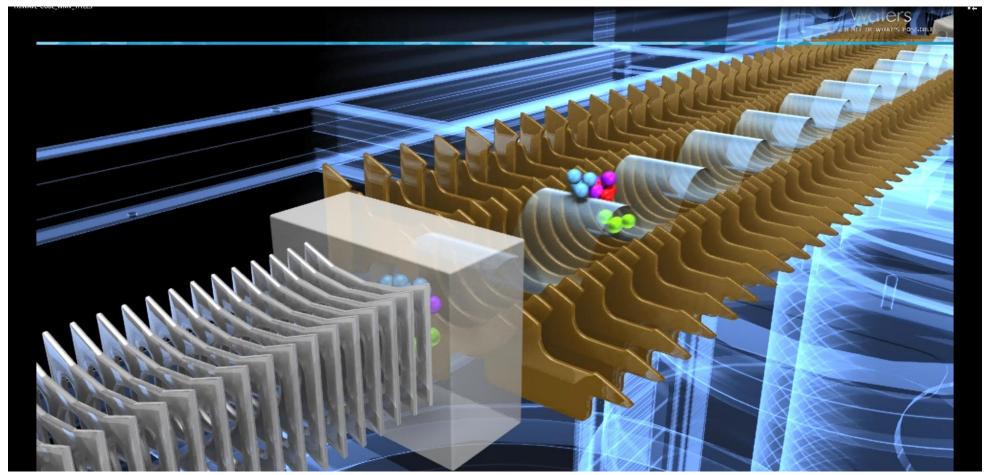
T-wave IMS

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Stacked Ring Ion Guide: Opposite phases of RF voltage are applied to adjacent electrodes to provide radial ion confinement and high transmission.

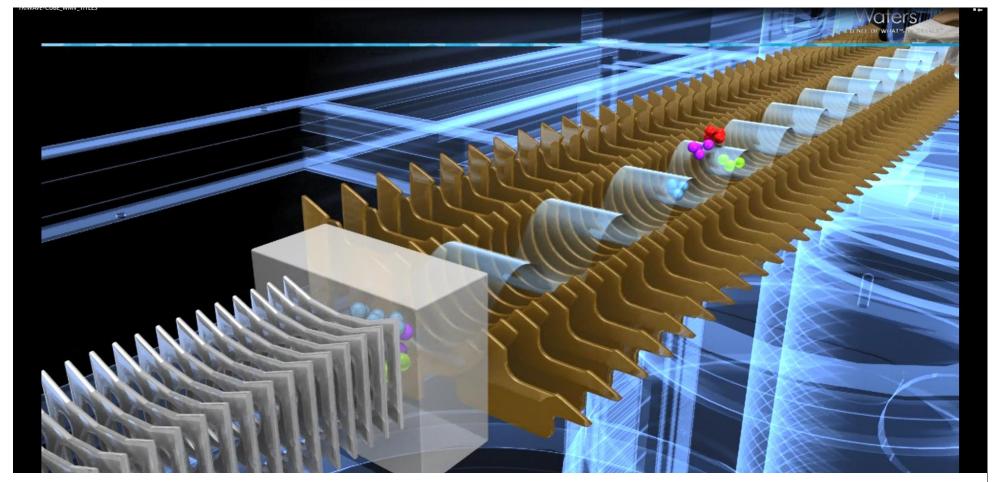
T-wave IMS Waters™



T-Wave DC voltages are applied to electrode pairs in repeating sequence along the device and step to respective adjacent electrode pairs.

T-wave IMS

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Ions are propelled along the ion mobility separation device.

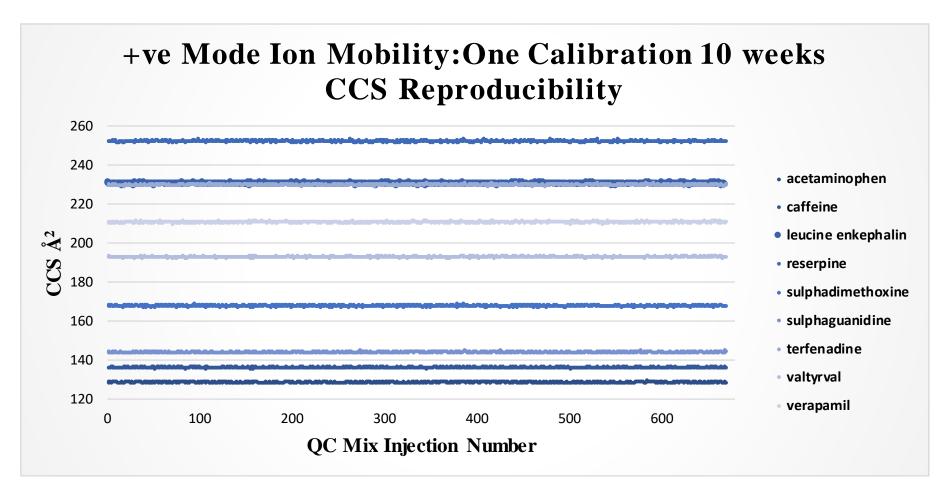
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CCS and Time Robustness

Robustness: CCS as a constant part 1



QC mix CCS reproducibility over 10 weeks, 24/7 acquisition ESI+ using a single ion mobility calibration.



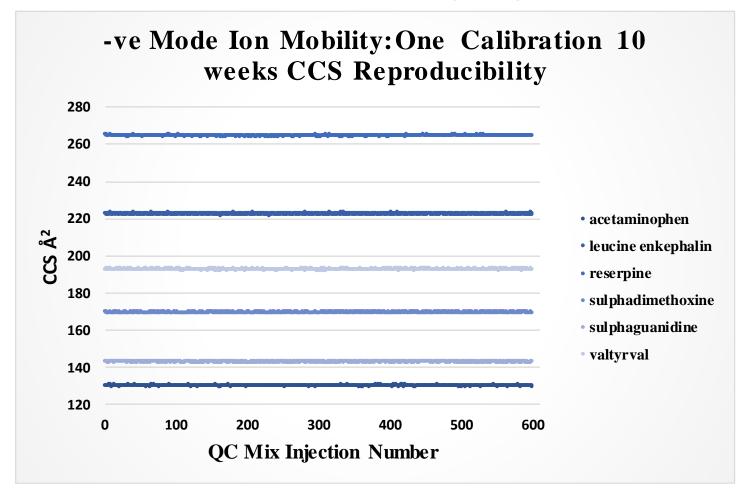
Waters Application Note March 2020 720006769EN AG-PDF: Small molecule ion mobility investigations into cross-platform and long-term robustness of a CCS metric Mike McCullagh¹; Michelle Wood²; Nayan Mistry²; Severine Goscinny³ and Petur Dalsgaard⁴

¹Waters Corporation, Wilmslow, United Kingdom; ²Sciensano, Brussels, Belgium; ³Department of Forensic Medicine, University of Copenhagen, Copenhagen, Denmark

Robustness: CCS as a constant part 2



QC mix CCS reproducibility over 10 weeks, 24/7 acquisition ESI- using a single ion mobility calibration.

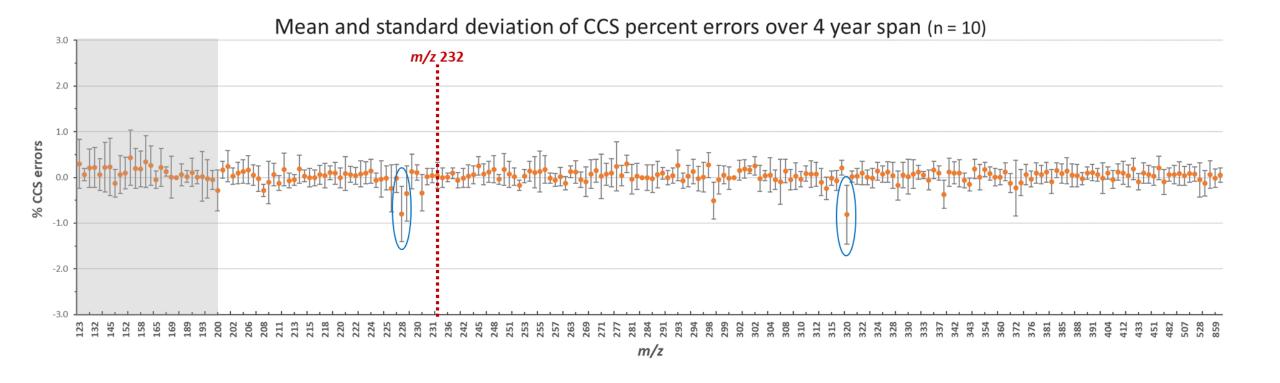


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Synapt mean CCS errors (±SD) for the 200 pesticides over 4 years. Zone in grey highlights the compounds with m/z below 200





Towards the use of ion mobility mass spectrometry-derived collision cross section as a screening approach for unambiguous identification of targeted pesticides in food. Séverine Goscinny, Michael McCullagh, Johann Far, Edwin De Pauw and Gauthier Eppe. *Rapid Commun Mass Spectrom*. 2019;1–15.

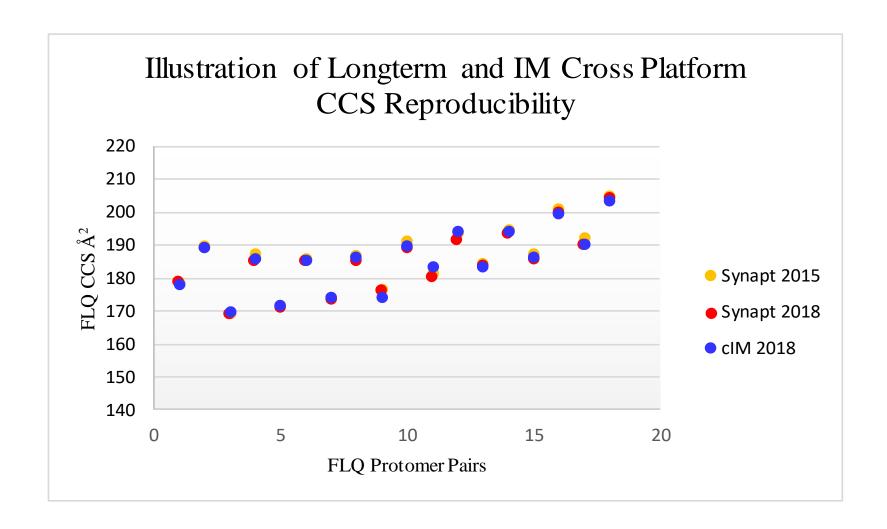
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CCS and Platform Robustness

Robustness between platforms (Waters): CCS as a constant part 1

Synapt linear IM (compared over a period of 3 years) and cIM for fluoroquinolone protomer CCS reproducibility.

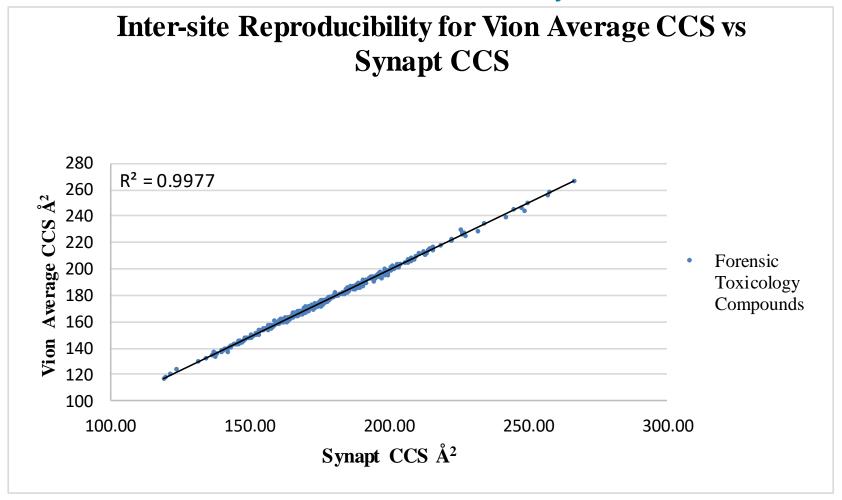




Investigations into the performance of travelling wave enabled conventional and cyclic ion mobility systems to characterise protomers of fluoroquinolone antibiotic residues Michael McCullagh, Kevin Gile Keith Richardson, Sara Stead and Martin Palmer. Rapid Commun Mass Spectrom. 2019;1–11

Vion inter-site CCS versus SYNAPT CCS and corresponding frequency distribution for inter-site Vion library Δ CCS.



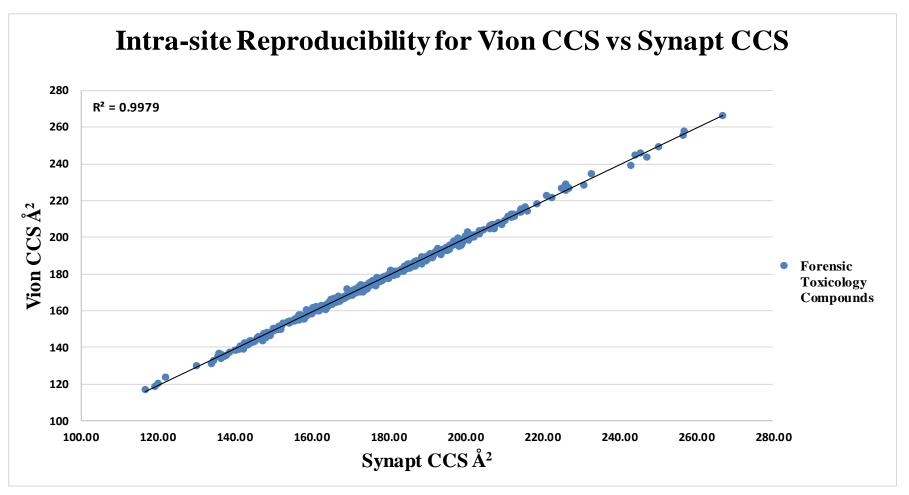


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Vion intra-site CCS versus SYNAPT CCS and corresponding frequency distribution for intra-site Vion library Δ CCS.



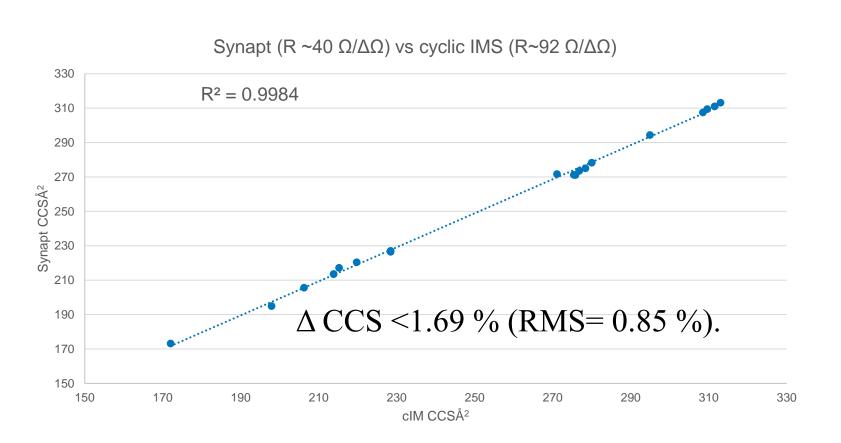


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Cross platform comparison of charged isomer species





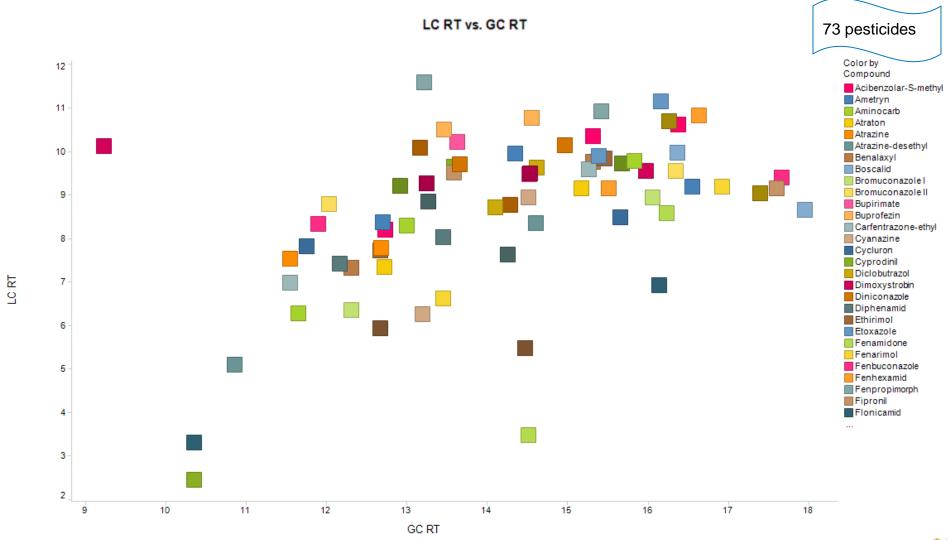
Metaflumizone (Z)-Isomer Species I Metaflumizone (Z)- Isomer Species II Metaflumizone (E)-Isomer Species I Metaflumizone (E)-Isomer Species II Spinosyn D [M+H]+ Spinosyn D Sodimer I Spinosyn D Sodimer II Spinosyn D Potassimer I Spinosyn D Potassimer II Spinosyn A [M+H]+ Spinosyn A Sodimer I Spinosyn A Sodimer II Spinosyn A Potassimer I Spinosyn A Potassimer II **Epoxiconazole** Spinosyn D Indoxacarb Protomer I Indoxacarb Protomer II Fenpyroximate Protomer I Fenpyroximate Protomer II

Avermectin B1a

Investigations into pesticide charge site isomers using conventional IM and cIM systems. Michael McCullagh, Séverine Goscinny, Martin Palmer, Jakub Ujma, Talanta, Volume 234, 2021, 122604.

GC and LC Retention Times

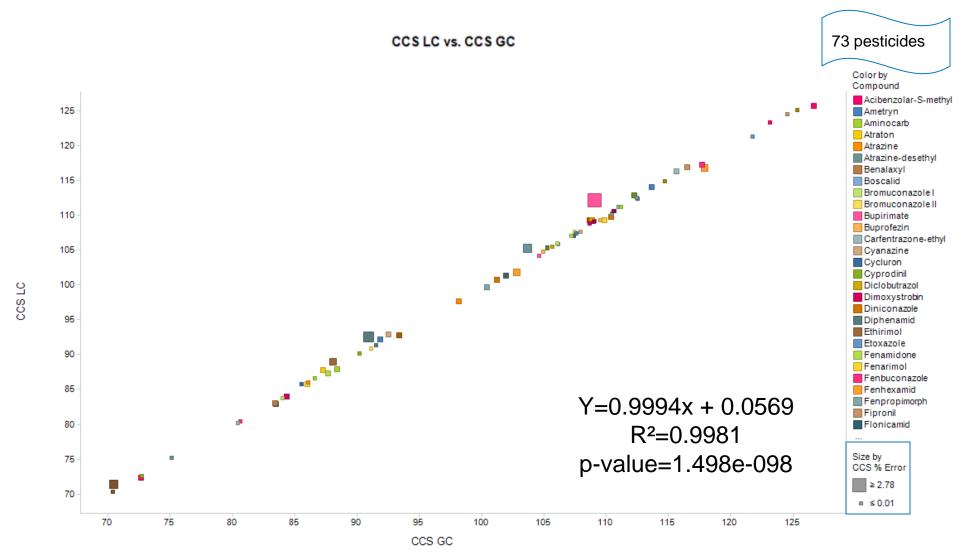
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GC and LC CCS Values: Independent of Inlet/Chromatography

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Time, Space and Platform Robustness and Indepence:

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- Within Waters portfolio : Yes
- Between Instrument Manufacturers? Maybe if ... $v_o = KE$

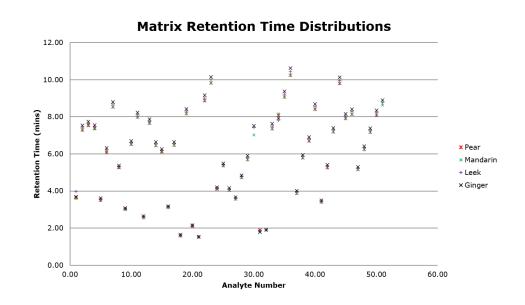
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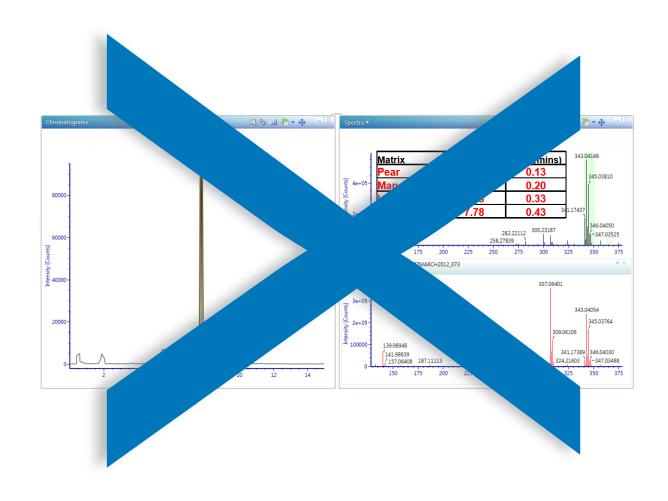
CCS and MATRIX ROBUSTNESS

Matrix and Chromatography (UHPLC)

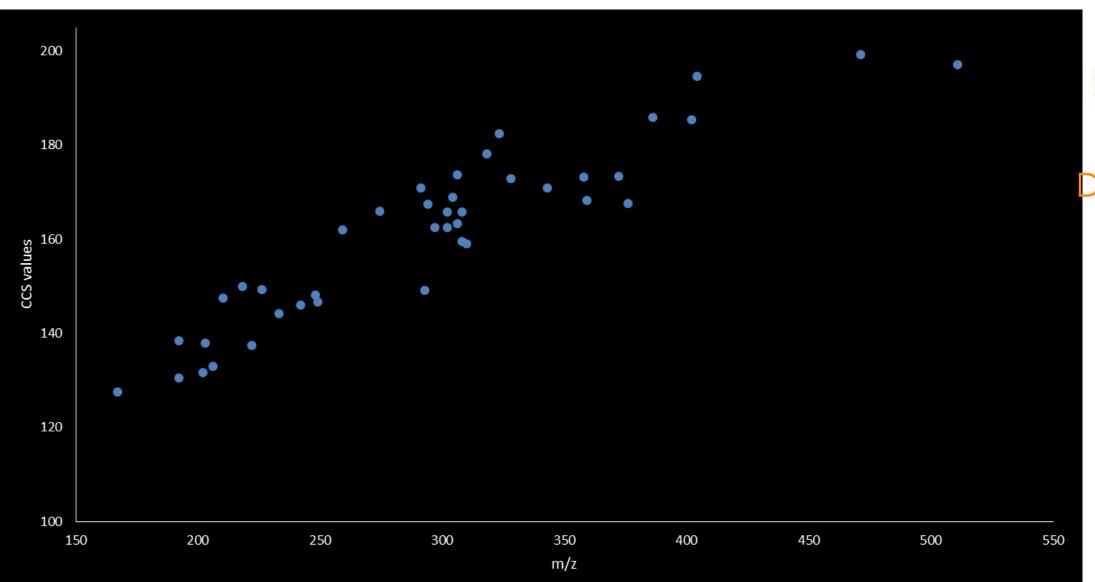
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Complex matrix dependent retention time shift: Boscalid as example



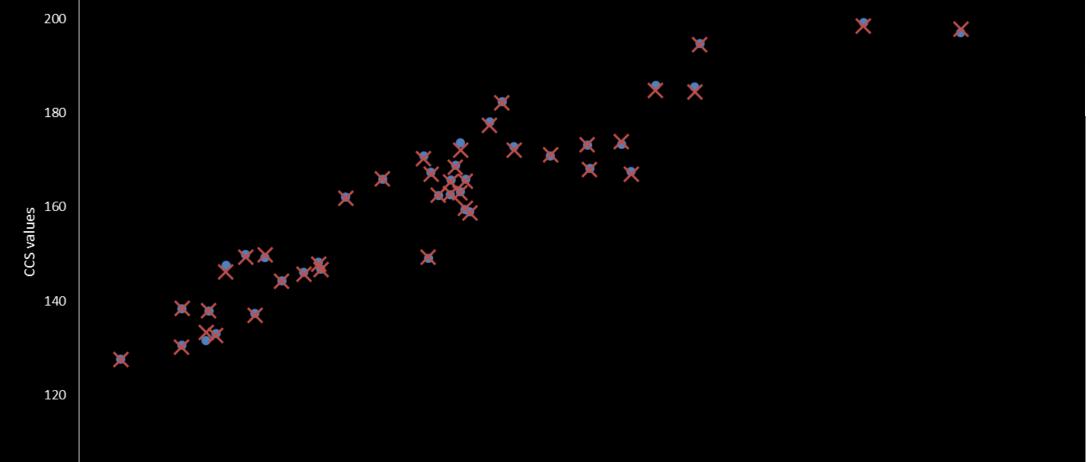








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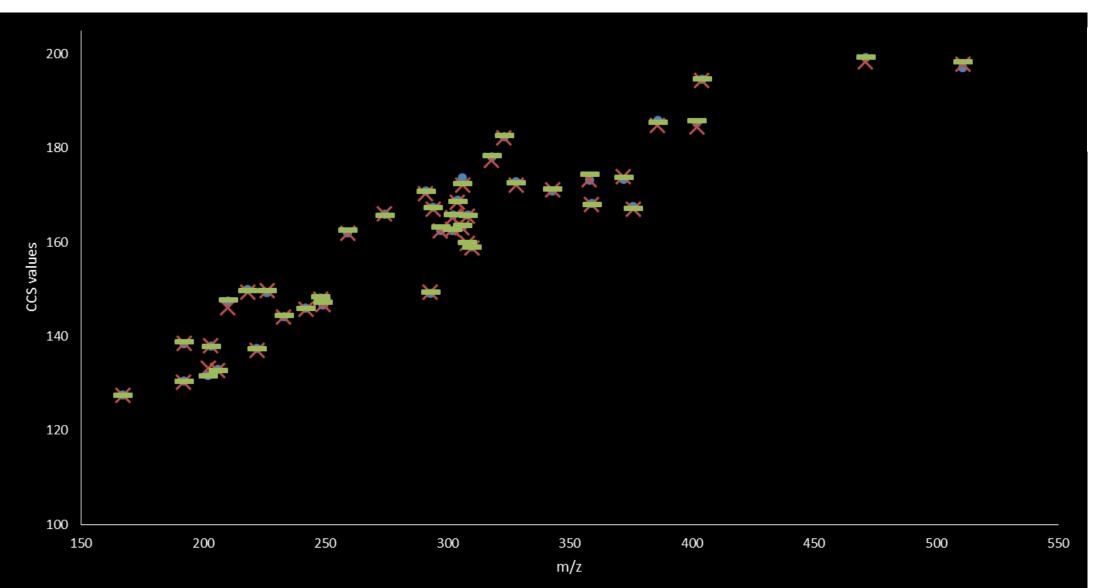


m/z



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Mixed rice

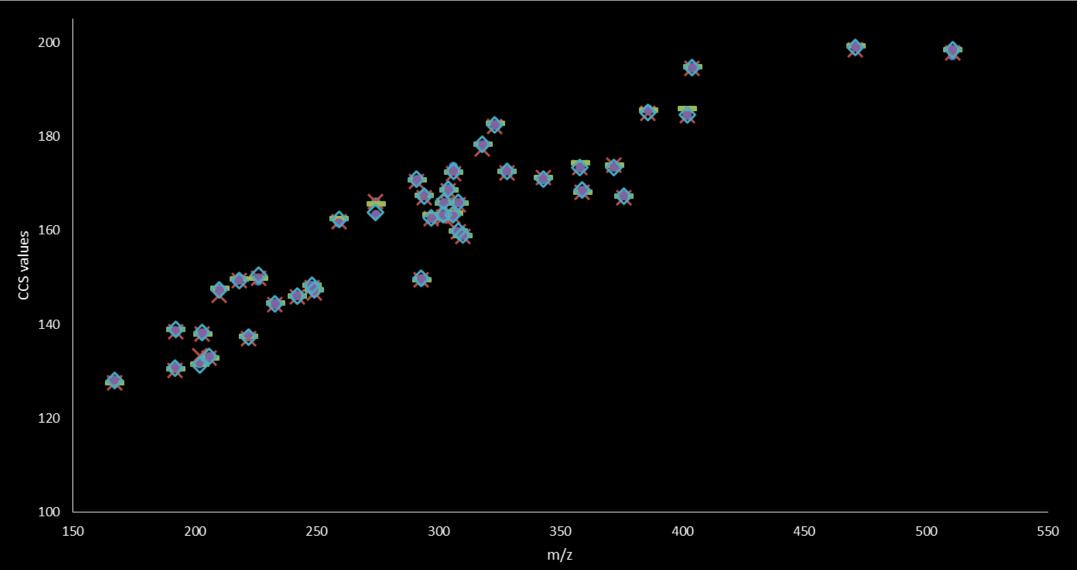
m/z

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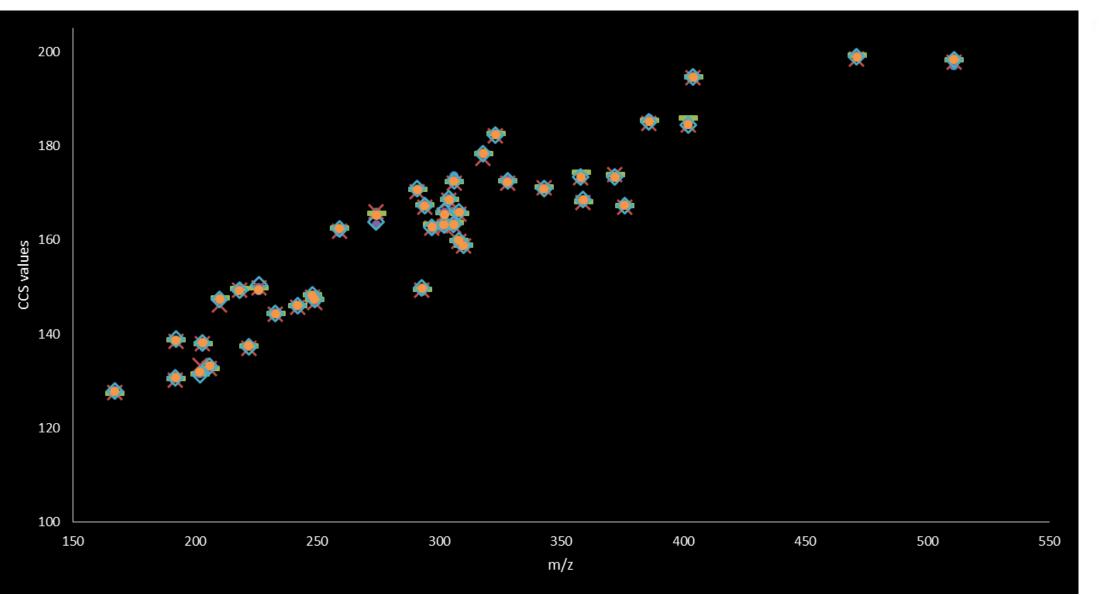




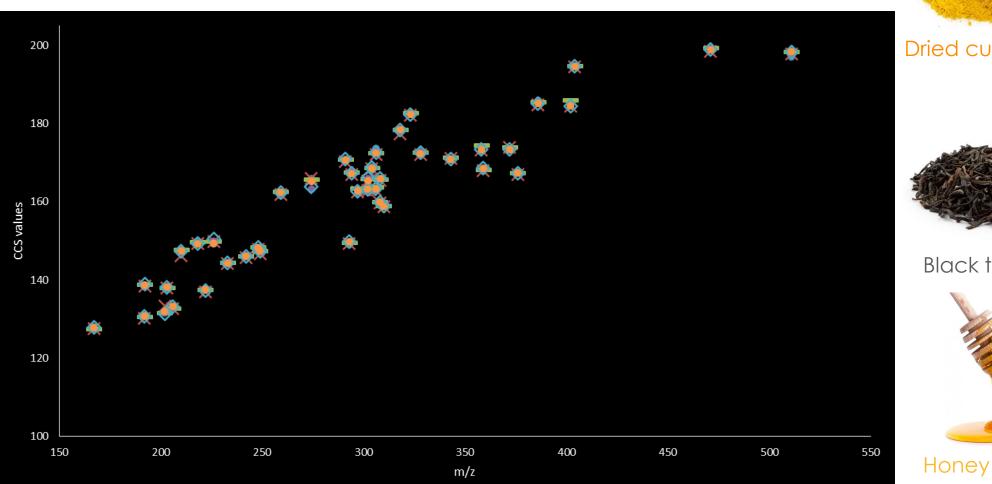




Apple



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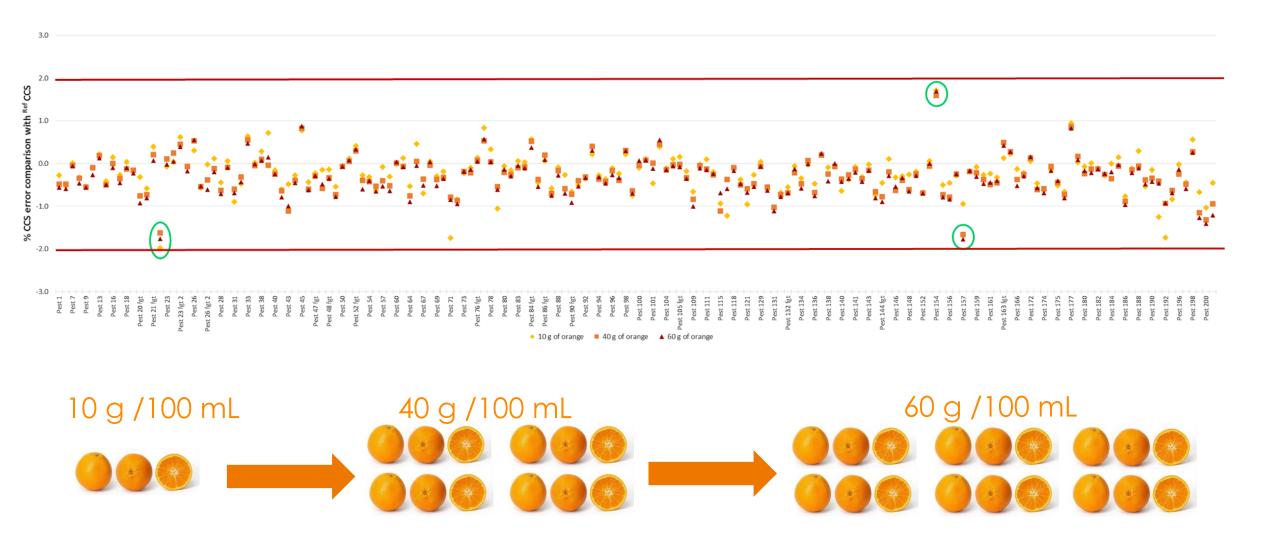




Courgette

Impact of matrix load on CCS measurement for pesticides at constant concentration (100 ng.mL⁻¹). Comparison shown as percent difference with RefCCS

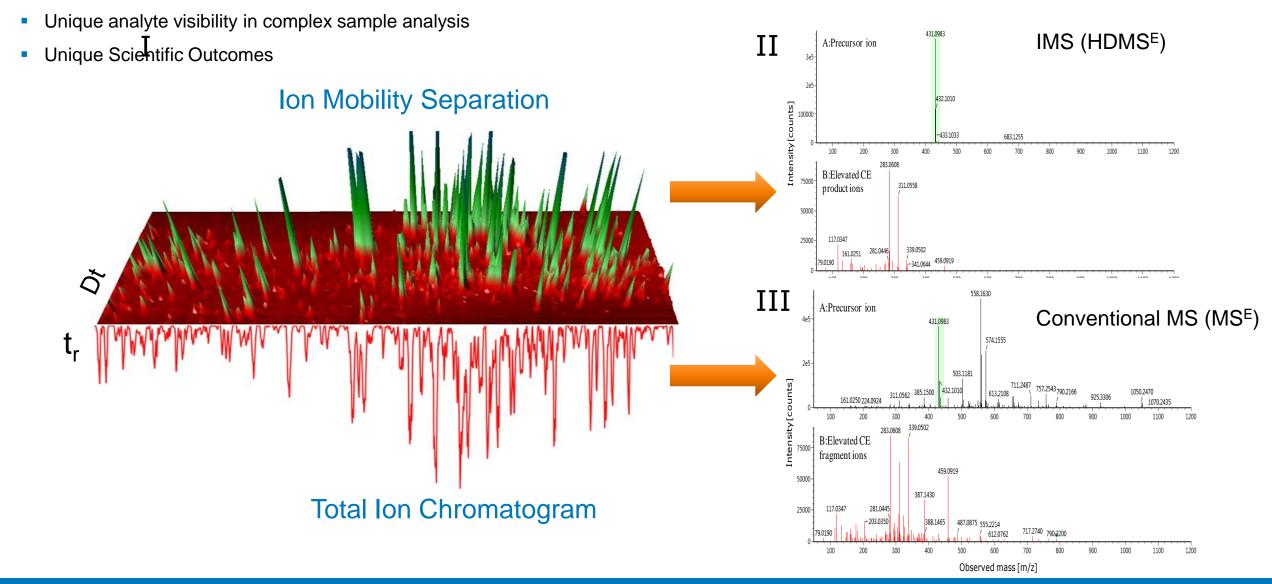




Towards the use of ion mobility mass spectrometry-derived collision cross section as a screening approach for unambiguous identification of targeted pesticides in food. Séverine Goscinny, Michael McCullagh, Johann Far, Edwin De Pauw and Gauthier Eppe. *Rapid Commun Mass Spectrom.* 2019;1–15.

UPLC Ion mobility mass spectrometry 3D resolution: enhanced peak capacity and selectivity





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Ion Mobility Examples: LC-IMS-HRMS

Steviol glycosides Analysis: Authentication Profiling Using CCS Libraries









What are Steviol glycosides.

Where/Why Are They Used.

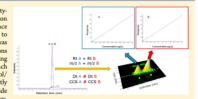
- Why is Analysis Required.
- Why IMS

Exploring the Complexity of Steviol Glycosides Analysis Using Ion Mobility Mass Spectrometry

Michael McCullagh, ** David Douce, Els Van Hoeck, and Severine Goscinny

Waters Corporation, Stamford Avenue, Altrincham Road, Wilmslow, SK9 4AX, U.K. *Scientific Institute of Public Health, Rue Juliette Wytsmans 14, 1050 Brussels, Belgium

ABSTRACT: A proof of principle method using ion mobilitymass spectrometry (IM-MS) and collision induced dissociation (CID) coupled with micro flow ultra high-performance chromatography (UHPLC-IM-MS) has been developed to screen for steviol glycosides. Traveling wave ion mobility was used to determine rotationally averaged collision cross sections in nitrogen buffer gas (TWCCSN2). To explore the evolving applicability of ion mobility screening, the analytical approach was initially developed and applied to the analysis of a steviol/ steviol glycoside spiked chocolate spread extract. Subsequently 55 food commodities were screened using a steviol glycoside TWCCSN₂ library. IM analyses produced TWCCSN₂ values,



enabling the unequivocal identification of the steviol glycosides and isomeric pairs (negating the reliance on product ions). In addition, coeluting isomeric species, comprising (labile fragment ions, doubly charged dimers, and multiply charged species) have been identified and resolved. Isomeric false detections were avoided, with the coeluting isomeric species quantified. A quantitative assessment of TWCCSN2 in the analysis of steviol glycosides was performed.

Production of food commodities containing sweeteners has increased; a driving force being to reduce global incidence of obesity and the associated health impact (such as diabetes type II).1 Sweeteners can provide reduced or zero calorific intake. They are a diverse group of chemical compounds derived from plants or chemical synthesis. As a food additive, safety evaluation of sweeteners is a basic requirement. The Joint FAO/WHO Expert Committee on Food Additives (JECFA) established regulations for steviol glycosides, requiring a purity level of at least 95% for seven chemically defined steviol glycosides.2 In Europe, this food additive (E 960) is authorized in specific food at defined rates (maximum permitted level) laid down in Commission Regulation No. 1131/2011,3 with acceptable purity criteria defined in Regulation (EU) No. 231/2012.4 This later legislation established for the pristine additive, a minimum of 95% content of 10 steviol glycosides with at least 75% of stevioside and/or rebaudioside A within

For consumers, the European Food Safety Authority (EFSA), perform risk assessments, and investigate health claims/beneficial effects related to sweeteners. A revised dietary exposure assessment of adults and children to steviol glycosides, through their use as a (food additive) was carried out. The variety of food products containing steviol glycosides as sweeteners in Europe is extensive. Smoked/dried fish, fruit based drinks, cocoa based confectionary, sweet/sour preserves, breakfast cereals, beers, ciders, sweeteners, and reduced sugar products are examples of where steviol glycosides exposure may occur in a European diet. Health effects of food additives/ sweeteners on consumers will be impacted by the level of dietary exposure, e.g., by work life balance, socioeconomics, lifestyle choice, and typical national cuisine. The mean dietary exposure to steviol glycosides is expressed in terms of steviol equivalents.5 The EFSA report details the revisions of acceptable use levels in a wide variety of food commodities and also the acceptable daily intake (ADI) of 4 mg/kg body weight (bw)/day for toddlers, which still exceeds the 95th percentile in a number of European Union countries.6

Reasons to explore a new approach for steviol glycoside analysis include meeting legislative requirements, (determination of ADI), ensuring all steviol glycosides (including isomers) are identified/detected and the true steviol equivalent determined. Also authentication profiling to determine origin in food commodities, optimization of stevia processing methodology, and breeding of Stevia rebaudiana Bertoni to produce more favorable flavor characteristics.7 In addition for product quality, where new information on steviol glycoside makeup could help characterize minor novel steviol glycosides, which may contribute to/reduce bitter aftertaste or have more potent sweetness intensity and can be characteristic of products containing steviol glycosides.^{8–10} Extracts of Stevia rebaudiana Bertoni leaves may contain isomers, which can have different

Received: December 1, 2017 Accepted: March 14, 2018 Published: March 14, 2018

ACS Publications © 2018 American Chemical Society

Safety of a proposed amendment of the specifications for steviol glycosides (E 960) as a food additive: to expand the list of steviol glycosides to all those identified in the leaves of Stevia Rebaudiana Bertoni

EFSA Panel on Food Additives and Flavourings (FAF), Maged Younes, Gabriele Aquilina, Karl-Heinz Engel, Paul Fowler, Maria Jose Frutos Fernandez, Peter Fürst, Rainer Gürtler, Ursula Gundert-Remy, Trine Husøy, Melania Manco, Wim Mennes, Peter Moldeus, Sabina Passamonti, Romina Shah, Ine Waalkens-Berendsen, Detlef Wölfle, Matthew Wright, Gisela Degen, Alessandra Giarola, Ana M Rincon and Laurence Castle

SCIENTIFIC OPINION ADOPTED: 24 March 2020

doi: 10.2903/j.efsa.2020.6106

The EFSA Panel on Food Additives and Flavourings (FAF) provides a scientific opinion on the safety of the proposed amendment of the specifications for stevial glycosides (E 960) as a food additive, in particular to expand the list of steviol glycosides to 60 steviol glycosides identified in the leaves of Stevia Rebaudiana Bertoni, With the existing specifications, the food additive must be comprised of not less than 95% of the 11 named steviol glycosides. The proposed change is to include all 60 steviol glycosides in the same limit value of 95% and this would allow the presence of up to 5% of impurities. FAF Panel considered that all steviol glycosides share the same metabolic fate, and therefore, the safety of 60 identified steviol glycosides can be based on read-across from toxicological data previously evaluated by EFSA and the acceptable daily intake (ADI) of 4 mg/kg body weight (bw) per day will apply to all those stevial alycosides. However, according to the proposed change in specifications, there remains a small but not insignificant fraction of the additive that would be undefined and therefore cannot be evaluated by the Panel. The Panel concluded that the inclusion of the 60 steviol divcosides in the proposed specifications for steviol glycoside (E960) would not be of safety concern. However, the Panel cannot condude on the safety of the proposed amendment to the specifications of steviol alvoosides (E 960) as food additive if the purity assay value of not less than 95% for the total content of steviol alvoosides is maintained.

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Keywords: Steviol glycosides, E960, food additive

Requestor: European Commission Question number: EFSA-Q-2019-00063 Correspondence: fip@efsa.europa.eu

www.efsa.europa.eu/efsajourna

FFSA Journal 2020:18/4):6106

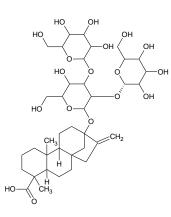


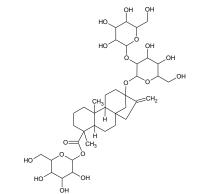


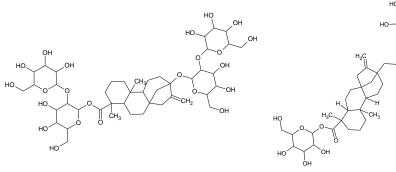
Steviol glycosides profiled using ion mobility

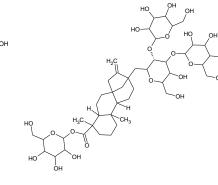
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$$H_3$$
C H_3 C H_4 H_5 C H_5 H_6 H_7 H_8 C H_8









Rubusoside M-H=641.3179 Da

Steviolbioside M-H=641.3179 Da

Rebaudioside B M-H=803.3707 Da

Stevioside M-H=803.3707 Da*

Rebaudioside E M-H=965.4235 Da

Rebaudioside A M-H=965.4235 Da*

Dulcoside A [M-H]⁻=787.3758 Da

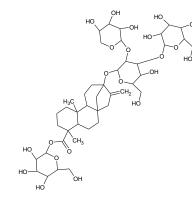
$$\begin{array}{c} \text{HO} \\ \text{OH} \\$$

Rebaudioside C $[M-H]^{-}=949.4286$ Da

$$\begin{array}{c} \text{CH}_3 \\ \text{O} \\ \text{CH}_3 \\ \text{OH} \end{array}$$

Steviol $[M-H]^{-}=317.2122$ Da

Rebaudioside D $[M-H]^{-}=1127.4763$ Da



Rebaudioside F $[M-H]^{-}=935.4129$ Da

Determined micro-flow UPLC retention times and TWCCSN₂ values

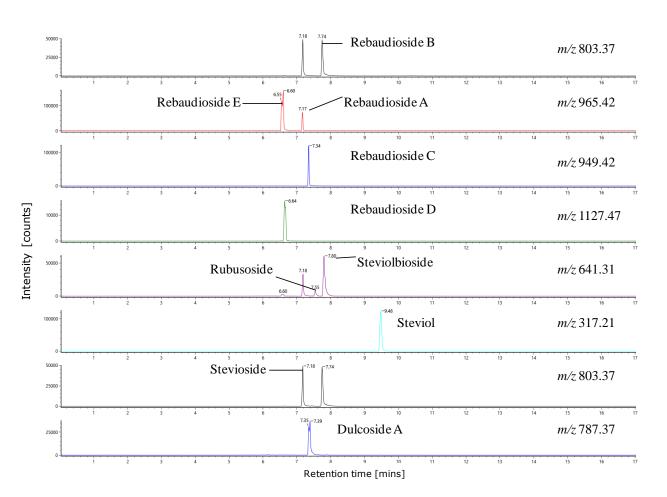


Compound	Formula	[M-H] ⁻ m/z	Expected Rt (mins)	Reference TWCCSN ₂ (Å ²)	Reference [M - H + HCO ₂ H] ⁻ ^{TW} CCSN ₂ (Å ²)
rebaudioside E	$C_{44}H_{70}O_{23}$	965.4230	6.61	289.2	298.7 (5%).
rebaudioside D	$C_{50}H_{80}O_{28} \ C_{44}H_{70}O_{23}$	1127.4758 965.4230	6.68 7.17	321.8	324.5 (11%) 311.3 (19%)
rebaudioside A stevioside	$C_{38}H_{60}O_{18}$	803.3701	7.20	298.9 269.6	278.1 (46%),
rebaudioside F	$egin{array}{c} ext{C}_{43} ext{H}_{68} ext{O}_{22} \ ext{C}_{44} ext{H}_{70} ext{O}_{22} \end{array}$	935.4124 949.4280	7.32 7.37	293.2	306.5 (9%), 308.1 (10%),
rebaudioside C	$C_{44}H_{70}O_{22}$ $C_{38}H_{60}O_{17}$	787.3752	7.37	299.5	308.1 (10%),
dulcoside A rubusoside	$C_{32}H_{50}O_{13}$	641.3173	7.56	270.6 241.3	250.2 (284%)
rebaudioside B	$C_{38}H_{60}O_{18}$	803.3701	7.77	261.2	
steviolbioside	$C_{32}H_{50}O_{13}$	641.3173	7.81	235.8	
steviol	$C_{20}H_{30}O_3$	317.2117	9.48	173.4	

Steviol glycoside characterisation (spiked chocolate spread extract matrix)

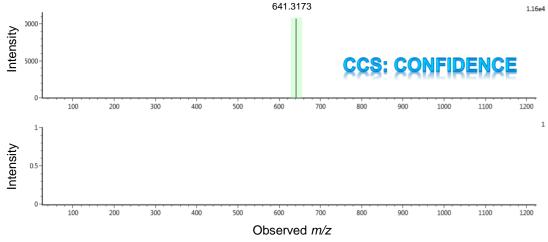
Extracted mass chromatograms for steviol and profiled steviol glycosides ≤100pg/µL spiked into chocolate spread extract.





Rebaudioside A 1pg
Dulcoside A 1pg
Rebaudioside F 0.76pg
Rebaudioside E 0.92pg
Rebaudioside B 0.7pg

Rebaudioside C 1pg Rebaudioside D 0.80pg Stevioside 1pg Rubusoside 0.68pg Steviolbioside 0.72pg

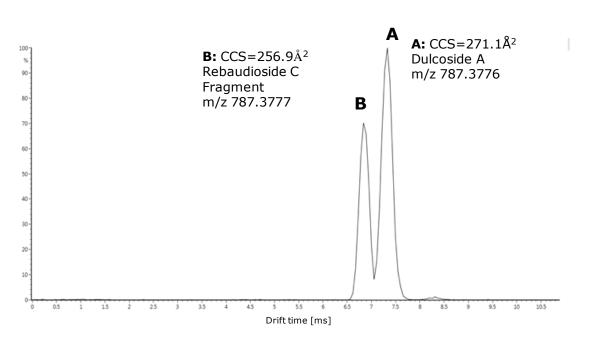


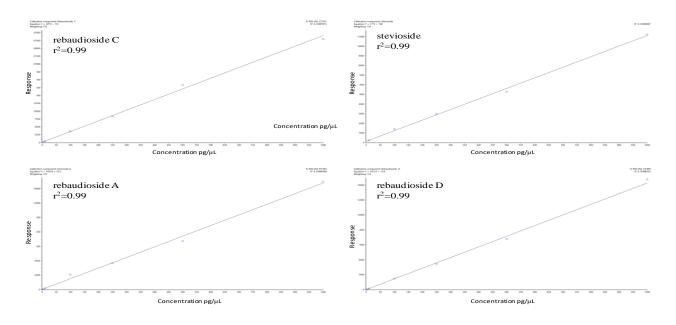
- Waters ACQUITY UPLC M-Class® System
- Miroflow reverse phase gradient: iKey @ 2.0µL/min
- Injection Volume: 1µI (full loop)

Chromatographically coeluting isomeric species observed

Ion mobility ATD drift time plot showing two mobility separated species at *m/z* 787.37 for Reb C and Dulcoside A







Ion Mobility convoluted isomeric quantitation of steviol glycosides



	Calculated Concentration pg/μL						
	HRMS ^E	HDMS ^E	HRMS ^E	HDMS ^E			
Sample code	dulcoside A	dulcoside A	stevioside	stevioside			
	[M-H] ⁻	[M-H] ⁻	[M-H] ⁻	[M-H] ⁻			
Sample 1	386.10	98.39*	1843.24	1374.0			
Sample 2	31.33	Not observed	1823.76	Not observed			
Sample 3	18.97	Not observed	1505.52	Not observed			
Sample 4	10.14	Not observed	1337.54	Not observed			
Sample 5	37.24	Not observed	1840.37	Not observed			
Sample 6	577.18	105.93	2315.34	2196.0			
Sample 7	128.54	Not observed	1620.93	1306.62			
Sample BE22	61.54	Not observed	1370.90	190.79			
Sample T9	503.15	262.58	2578.72	2285.08			
Sample TB6	152.90	Not observed	1730.29	1610.28			
Sample BB1	549.45	Not observed	1682.17	1404.13			

Soft drink analysis 2020: And still working in 2023!

Waters™

- Robust, reproducible CCS values: Standing the test of time
- Methodology: UPLC-IM-MS
- Steviol glycosides identified using CCS: Retention times determined using 10 min retention time window

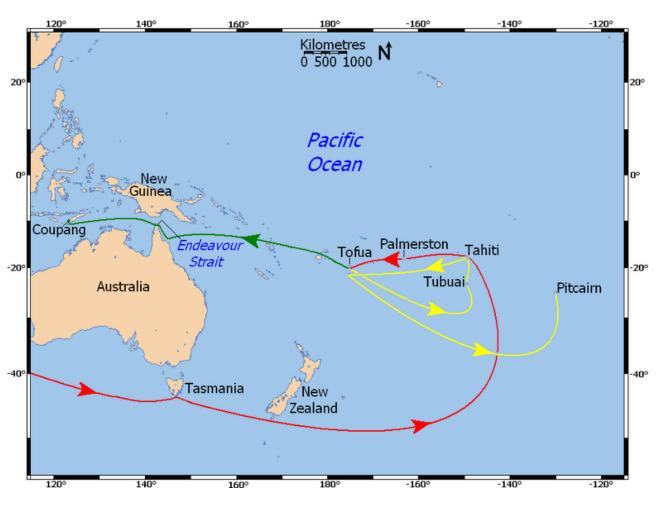
Component Summary ▼											
Component name	Identification status	Observed m/z	Mass error (ppm)	Expected RT (min)	Observed RT (min)	Observed CCS (Ų)	Expected CCS (Ų)	CCS delta (%)	Response	Expected Fragments Found	Adducts
1 Citric acid	Identified	191.0193	-2.1	0.79	0.76	127.41	127.30	0.08	54912	0	-H
Dulcoside A	Identified	787.3749	-1.2	5.06	5.30	269.53	270.05	-0.19	282	0	-H
3 Dulcoside A Adduct	Identified	833.3828	1.9	5.06	5.06	277.10	275.23	0.68	3078	0	+HCOO
4 Hesperidin 1	Identified	609.1818	-1.1	3.44	3.48	231.22	231.15	0.03	5988	1	-H
Hesperidin 2	Identified	609.1818	-1.1	3.44	3.48	242.90	242.96	-0.03	2010	1	-H
⁶ Rebaudioside A	Identified	965.4240	0.5	4.81	4.81	296.21	298.90	-0.90	134205	1	-H
⁷ Rebaudioside B	Identified	803.3725	2.3	5.38	5.39	262.08	261.20	0.34	31245	0	-H
⁸ Rebaudioside C	Identified	949.4308	2.3	5.01	5.01	300.12	299.50	0.21	74692	0	-H
⁹ Rebaudioside D	Identified	1127.4765	0.2	4.17	4.17	323.68	321.80	0.58	2264	0	-H
Rebaudioside F	Identified	935.4145	1.7	4.95	4.96	293.77	293.20	0.19	32357	1	-H
Rebaudioside A Fragm	Identified	803.3690	-2.1	4.81	4.81	261.15	261.20	-0.02	123587	0	-H
Rebaudioside C Fragm	Identified	787.3754	-0.5	5.01	5.01	257.80	256.95	0.33	40854	0	-H
Rebaudioside E	Identified	965.4243	0.8	4.09	4.09	290.46	289.20	0.44	830	0	-H
Rubusoside	Identified	641.3165	-2.1	5.25	5.25	242.69	241.30	0.57	2862	0	-H
Rubusoside adduct	Identified	687.3237	0.5	5.25	5.25	251.08	249.63	0.58	7761	0	+HCOO
16 Stevioside	Identified	803.3692	-1.9	4.85	4.85	270.86	269.60	0.47	99541	0	-H

	Column temp 45 °C								
ļ	C18 HSS T3 2.1x 100 mm								
	Time	Flow	%A	%В	Curve				
	0	0.4	95	5	6				
	0.5	0.4	95	5	6				
	6	0.4	0	100	6				
	9	0.4	0	100	6				
	9.5	0.4	95	5	6				
	11	0.4	95	5	6				

The "HMS Bounty Rule" Never Trade Dimensions for Accuracy (Bleiner et al. (in Prep))

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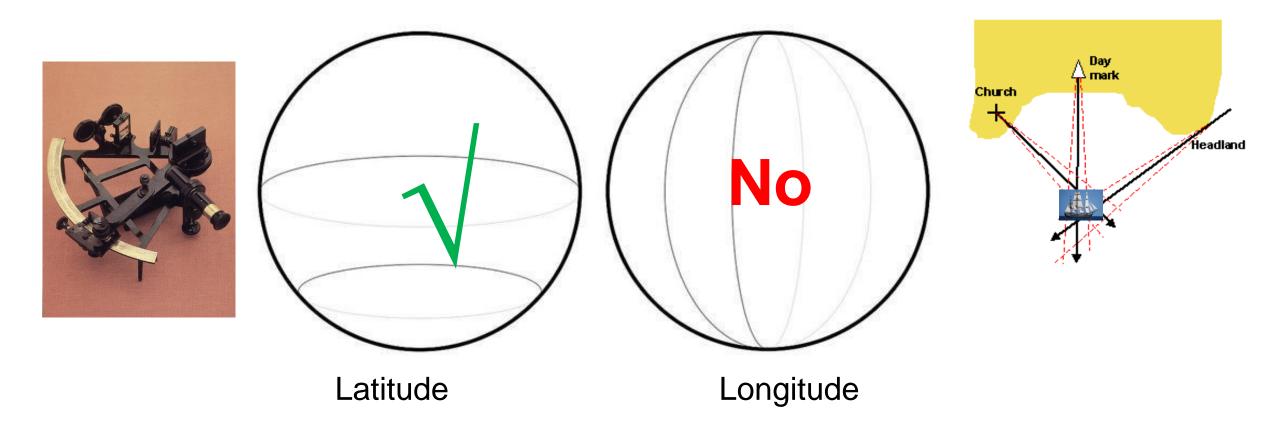




Wikipedia 2023

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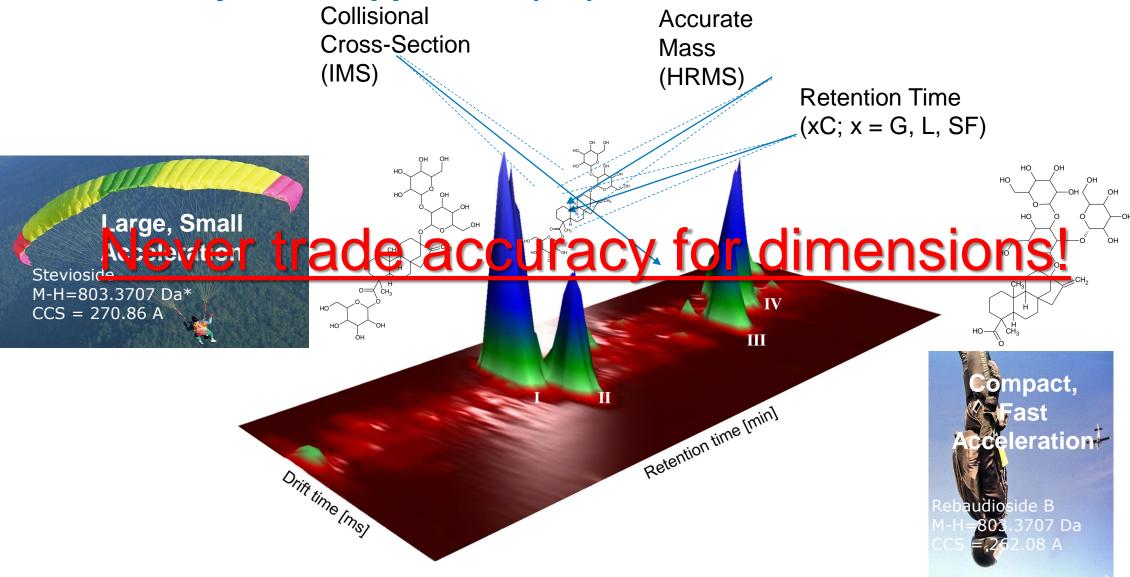
Before accurate chronometers: No accurate determination of Longitude



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"The HMS Bounty Rule" applied to (LC) HRMS

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Adapted from M. McCullagh et al 2021

Collaboration between Merck KGaA and Waters

Waters™

Home > Press Releases > Merck and Waters to Collaborate on Extractables and Leachables Reference Library

Merck and Waters to Collaborate on Extractables and Leachables Reference Library

- Library will enable analytical labs to identify potential E&L compounds in their samples using instruments from Waters and confirm
 the identity using Merck's reference materials
- Collaboration to provide testing labs with unrivalled confidence in their results and enhance consumer safety

Darmstadt, Germany, February 15, 2022 – Merck, a leading science and technology company, today announced that its Life Science business sector has entered into a collaboration with Waters Corporation to build and expand an Extractables and Leachables (E&L) Reference Library to include ion mobility measurements. The library will enable analytical labs to identify potential extractables and leachable compounds in their samples by using Waters' ion mobility-enabled Liquid Chromatography Mass Spectrometry (LC-MS) instruments and then confirming the identity and quantity using Merck reference materials.



"Accurate screening for extractables and leachables is imperative to ensuring consumer safety, especially in pharmaceuticals, food packaging, or

medical devices," said Heike Petri, Head of Advanced Analytical and Industrial & Testing, "This collaboration will provide manufacturers with unrivalled confidence in their results, help improve workflow efficiency for labs, and ultimately contribute to consumer safety."

Under the agreement, Waters will use high-quality analytical standards and Reference Materials from Merck to build and expand an E&L library of collision cross-section (CCS) values for Waters' LC-MS instruments. The library, which will be available for download from the Waters Marketplace (login required), will help to identify E&L compounds, with each addition to the library carefully selected to ensure maximum relevance to users. The library is cross-linked to the Merck online product catalogue to provide users access to reference materials to confirm their results.

About Merck

Merck, a leading science and technology company, operates across healthcare, life science, and electronics. Around 58,000 employees work to make a positive difference to millions of people's lives every day by creating more joyful and sustainable ways to live. From advancing gene-editing technologies and discovering unique ways to treat the most challenging diseases to enabling the intelligence of devices – the company is everywhere. In 2020, Merck generated sales of € 17.5 billion in 66 countries.

Scientific exploration and responsible entrepreneurship have been key to Merck's technological and scientific advances. This is how Merck has thrived since its founding in 1668. The founding family remains the majority owner of the publicly listed company. Merck holds the global rights to the Merck name and brand. The only exceptions are the United States and Canada, where the business sectors of Merck operate as EMD Serono in healthcare, MilliporeSigma in life science, and EMD Electronics.

All Merck news releases are distributed by email at the same time they become available on the Merck website. Please go to www.merckgroup.com/subscribe to register online, change your selection or discontinue this service.

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timo.breiner@merckgroup.com

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Collaborating with MilliporeSigma to incorporate CCS values

Accurate screening for extractables and leachables (E&Ls) is imperative for consumer safety, no matter the product. Be it a pharmaceutical, an item of food packaging, or a medical device, consumers place their trust in manufacturers to protect them from the thousands of potentially harmful chemicals that can leach into their products.

Through an exciting collaboration between Waters Corporation and the Life Science Business of MilliporeSigma, cross-collision (CCS) values are referenced on Sigma-Aldrich.com for LC/MS amendable E&L Certified Reference Materials (CRM) and analytical standards. As ion mobility spectrometry (IMS) becomes more routine, CCS values provide an additional identification point, strengthening results for labs and enhancing safety for consumers.

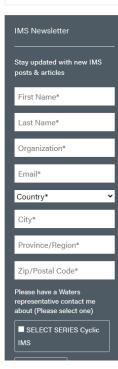
E&L Reference Materials that you can trust

As part of the collaboration, Waters and MilliporeSigma have jointly constructed a comprehensive UPLC-IMS-MS E&L library based on high-quality reference materials. The library can help users screen for a wide array of E&L compounds, with each addition to the library carefully selected to ensure maximum relevance to users. Each reference material also provides users with major identification points including m/z values for product ions, fragment ions and CCS.

Through application of IMS, the rotationally-averaged collision cross-section (CCS) of a given ion can be determined. It is a distinguishing characteristic and provides users with an additional descriptor that can improve confidence in their results.

We sat down with some of the minds that helped drive this collaboration forward so they could share their insights into how it will support analytical labs and enable greater throughput efficiency. Speaking to us from the Waters team were Mike McCullagh, [Consultant Scientist], Ben MacCreath, [Chemical Materials Business Operations], and Jens Jacobsen, [Sales Specialist] at Waters. From the MilliporeSigma side, we spoke to Coralie Leonard, [Business development, licensing and innovation manager], Dr Matthias Nold, [Product Manager for Reference Materials], and Markus Obkircher, [Director of R&D].

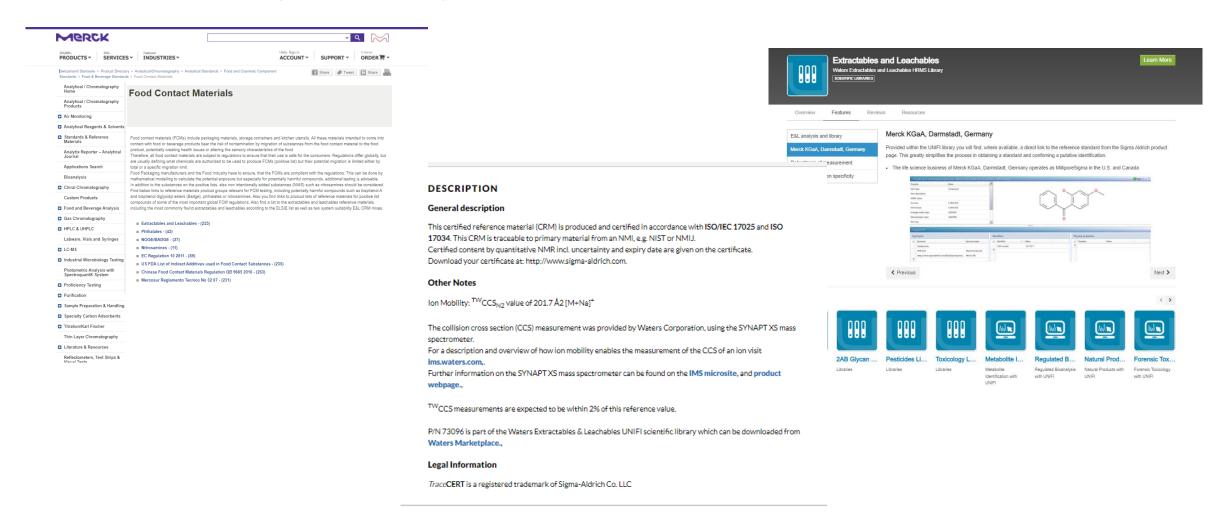
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Collaboration with Sigma-Aldrich (Merck KGaA) part 2:

Cross referencing Merck (Sigma-Aldrich) and Waters





https://www.sigmaaldrich.com/DE/en/product/sial/73096?context=product

Summary

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- lacktriangle CCS is complementary to mass and retention time (3 dimensions of resolution) $\sqrt{}$
- CCS robustness√
- Non-targeted screening in complex matrices√
- Reduced false detection rates √
- Differentiated isomers using CCS√
- Low intensity monoisotopic peak/ no product ions: CCS additional metric $\sqrt{}$
- Non labile compounds/ no product ions: CCS additional metric $\sqrt{}$
- Unique isomeric quantitation: CCS additional metric $\sqrt{}$
- True calculated concentration determined $\sqrt{}$
- Adducts provide an additional descriptor $\sqrt{}$
- Ion mobility product ions $\sqrt{}$

Acknowledgements

Waters™

- Eurachem for inviting ©
- Mike McCullagh Waters & Waters Corp Wilmslow
- Markus Obkircher, Sigma-Aldrich (Merck) Switzerland
- Matthias Nold & Hanspeter Sprecher Sigma-Aldrich (Merck) Switzerland
- Coralie Leonard, Sigma-Aldrich (Merck), Europe



Recent Publications



- Use of ion mobility mass spectrometry to enhance cumulative analytical specificity and separation to profile 6-C/8-Cglycosylflavone critical isomer pairs and known-unknowns in medicinal plants. Michael McCullagh, Cintia Alessandra Matiucci Pereira and Janete Harumi Yariwake. Phytochemical Analysis. 2019;1–13
- Exploring the Complexity of Steviol Glycosides Analysis Using Ion Mobility Mass Spectrometry. Michael McCullagh, David Douce, Els Van Hoeck and Severine Goscinny. Anal. Chem. 2018, 90, 4585–4595.
- Investigations into the performance of travelling wave enabled conventional and cyclic ion mobility systems to characterise protomers of fluoroquinolone antibiotic residues. Michael McCullagh, Kevin Giles, Keith Richardson, Sara Stead and Martin Palmer. Rapid Commun Mass Spectrom. 2019;1–11.
- Towards the use of ion mobility mass spectrometry-derived collision cross section as a screening approach for unambiguous identification of targeted pesticides in food. Séverine Goscinny, Michael McCullagh, Johann Far, Edwin De Pauw and Gauthier Eppe. Rapid Commun Mass Spectrom. 2019;1–15.
- A Comparison of Collision Cross Section Values Obtained via Ion Mobility Spectrometry Following Direct Infusion and an Evaluation of U(H)PLC-IM-MS for the Characterisation of Metabolites in Rat Urine. Leanne C. Nye, Jonathan P. Williams, Nyasha C. Munjoma, Marine P.M. Letertre, Muireann Coen, Robbin Bouwmeester, Jeremy K. Nicholson, Robert S. Plumb, Michael McCullagh, Lee A, Gethings, Steven Lai, James I. Langridge, Johannes P.C. Vissers, Ian D. Wilson. J Chromatogr A2019 Sep 27;1602:386-396.

Recent Publications



- **Profiling of the known-unknown** *Passiflora* **variant complement by liquid chromatography Ion mobility Mass spectrometry.** Michael McCullagh, Jeff Goshawk, David Eatough, Russell J. Mortishire-Smith, Cintia AM. Pereira, Janete H. Yariwake, Johannes PC. Vissers. Talanta 221 (2021) 121311
- Travelling Wave Ion Mobility-Derived Collision Cross Section for Mycotoxins: investigating interlaboratory and interplatform reproducibility. Laura Righetti, Nicola Dreolin, Alberto Celma, Mike McCullagh, Gitte Barknowitz, Juan V. Sancho, Chiara Dall'Asta J. Agric. Food Chem. 2020, Publication Date: September 1, 2020.
- APPLICATION NOTES
- Collision Cross Section- A New Identification Point for a "Catch All" Non-Targeted Screening Approach. Michael McCullagh¹ and Severine Goscinny². ¹Waters Corporation, Wilmslow, United Kingdom; ²Sciensano, Brussels, Belgium. Waters Application Note June 2014 720005055EN.
- **Discovery of pesticide protomers using routine ion mobility screening**. Michael McCullagh and Severine Goscinny^{2, 1}Waters Corporation, Wilmslow, United Kingdom; ²Sciensano, Brussels, Belgium. *Note* 2014, 720005028E.
- Ion Mobility in a Routine Workflow to Understand the Challenge of Analyzing Fluoroquinolone Antibiotic Residues. Michael McCullagh and Sara Stead. Waters Corporation, Wilmslow, United Kingdom Note June 2014, 720005078E.
- Cleanup and Collision Cross Section Values tolncrease Confidence and Efficiency in Pesticide Residues Screening Strategies. Michael McCullagh and Severine Goscinny^{2. 1}Waters Corporation, Wilmslow, United Kingdom; ²Sciensano, Brussels, Belgium. *Note* 2014, 720005080E.
- Enhancing analysis specificity and deconvolution of natural products using a positive mode ion mobility mass spectrometry library. Michael McCullagh; Jeff Goshawk; Russell Mortishire-Smith. Waters Application Note September 2019 720006650EN
- Un-paralleled multi-factor authentication mass spectrometry of complex natural products utilising UPLC and ion mobility. Michael McCullagh; Russell Mortishire-Smith and Jeff Goshawk. Waters Application Note March 2020 720006791EN
- Small molecule ion mobility investigations into cross-platform and long-term robustness of a CCS metric. Mike McCullagh¹; Michelle Wood²; Nayan Mistry²; Severine Goscinny³ and Petur Dalsgaard⁴. ¹Waters Corporation, Wilmslow, United Kingdom; ²Sciensano, Brussels, Belgium; ³Department of Forensic Medicine, University of Copenhagen, Copenhagen, Denmark. Waters Application Note March 2020 720006769EN.
- Use of ion mobility ^{TW}CCSN₂ values in non-targeted food additives screening Michael McCullagh¹; Jeff Goshawk¹, and Severine Goscinny². ¹Waters Corporation, Wilmslow, United Kingdom; ²Sciensano, Brussels, Belgium. Waters Application Note February 2020 720006768EN.
- A Workflow for Automatic MS Library Creation from Time-of-Flight Full-Spectra Data Processed in UNIFI. Jeff Goshawk, Gitte Barknowitz, and Michael McCullagh February 2020 720006783EN