

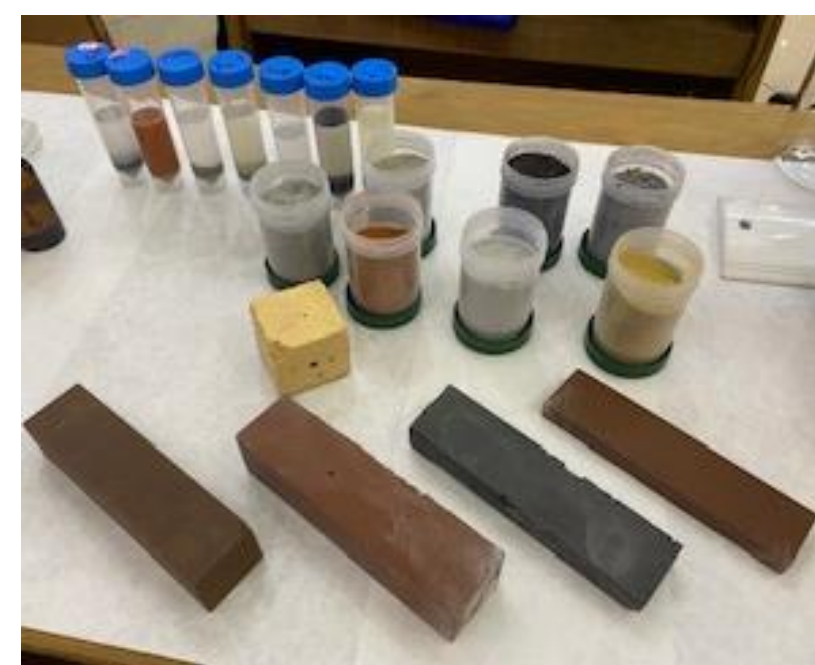
# Ensuring quality of the analytical process in a research laboratory

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**Abstract:** Validation and verification of analytical methods for testing of new materials or new specific methods often pose a line of problems and require alternative approaches. The necessary analytical instrumentation, the available CRM or standard procedures are among the problems encountered. Firstly, in the research laboratory usually the equipment is tailored to the quantity and size of the samples tested. Often we need to scale down the procedure to lower weight or volume, which could alter the analytical behavior of the standard procedure used for CRM testing. Secondly, although we chose the CRM with characteristics as close as possible to the studied material, often an appropriate CRM is not commercially available. For example, analysing CRM of soils is usually a procedure for estimating the analytical behaviour of the applied method for testing mine tailings, however the discrepancy of the matrix composition or analytes content, as well as their state and speciation in the sample, doesn't allow the method to be fully studied. In some cases, standard addition approach could help for at least approximate estimation of the uncertainty of measurement, but it is difficult to attend chemical equilibrium of added analyte with matrix to the level that reflects its state in the CRM sample or testing samples, especially in solid samples. Thirdly, when studying the behaviour of newly developed procedure for estimation of specific characteristics of the sample (for example, study of alkaline reactivity of mine tailing or fly ashes as precursors for geopolymer obtaining) the problems are even more pronounced. Applying the procedure to available CRM of soils or mine tailing doesn't provide the informative data as the reference or consensus values are not available for this specific procedure. An alternative approach was to study the behaviour of pure components and phases of the sample in the tested media. However, we found that we are scaling up the method and the obtained data could only roughly present the behaviour of the method in the tested materials. Fourthly, to ensure the quality control, CRM and in-lab control samples were used in our laboratory. However, we found that the differences in particle size in standard and testing samples highly influenced the obtained results and sometimes they appeared non-informative to be applied for quality control. In the research laboratory typically one analyses limited number different samples in different conditions applying modified procedures which limits the obtaining of reasonable quantity of data for construction of useful control charts. Nevertheless the mentioned problems we are still applying the standard procedures to available CRM for initial estimation of the uncertainty of measurement and the method behaviour in the research laboratory. More thorough estimation needs development of specific protocols.

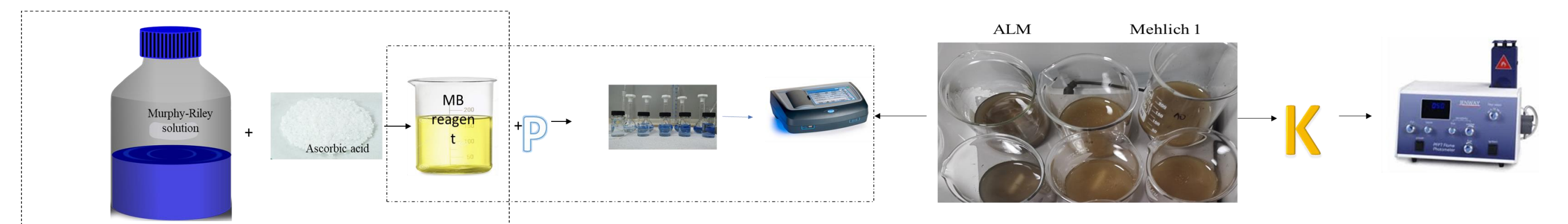
## Case 1: Scaling down the SOP - sequential extraction of mine tailing and fly ash



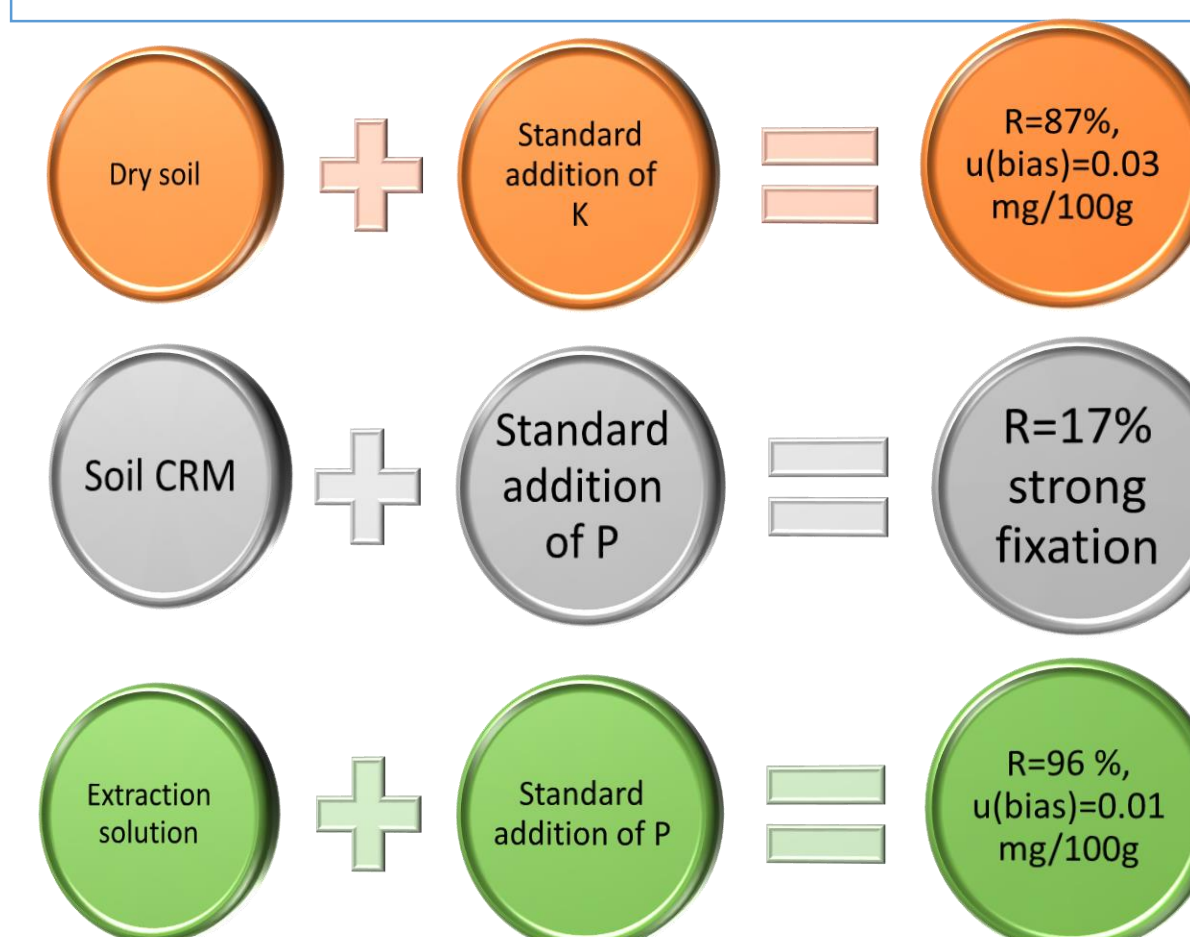
In the research laboratory usually the equipment is tailored to the quantity and size of the samples tested. Often we need to scale down the procedure to lower weight or volume, which could alter the analytical behavior of the standard procedure used for CRM testing.

## Case 2: Sample and CRM - standard addition

Although we chose the CRM with characteristics as close as possible to the studied material, often an appropriate CRM is not commercially available. For example, analysing CRM of soils is usually a procedure for estimating the analytical behaviour of the applied method for testing mine tailings, however the discrepancy of the matrix composition or analytes content, as well as their state and speciation in the sample, doesn't allow the method to be fully studied. In some cases, standard addition approach could help for at least approximate estimation of the uncertainty of measurement, but it is difficult to attend chemical equilibrium of added analyte with matrix to the level that reflects its state in the CRM sample or testing samples, especially in solid samples.



**Measurands:** extractable (Plant available) potassium and phosphorus in arable soils after lactate-acetate buffers extraction. Two variants of the standard addition approach: (1) to assess method bias – the standard addition of potassium CRM solution before extraction and (2) to assess measurement bias – standard addition potassium CRM after extraction and before measurement.



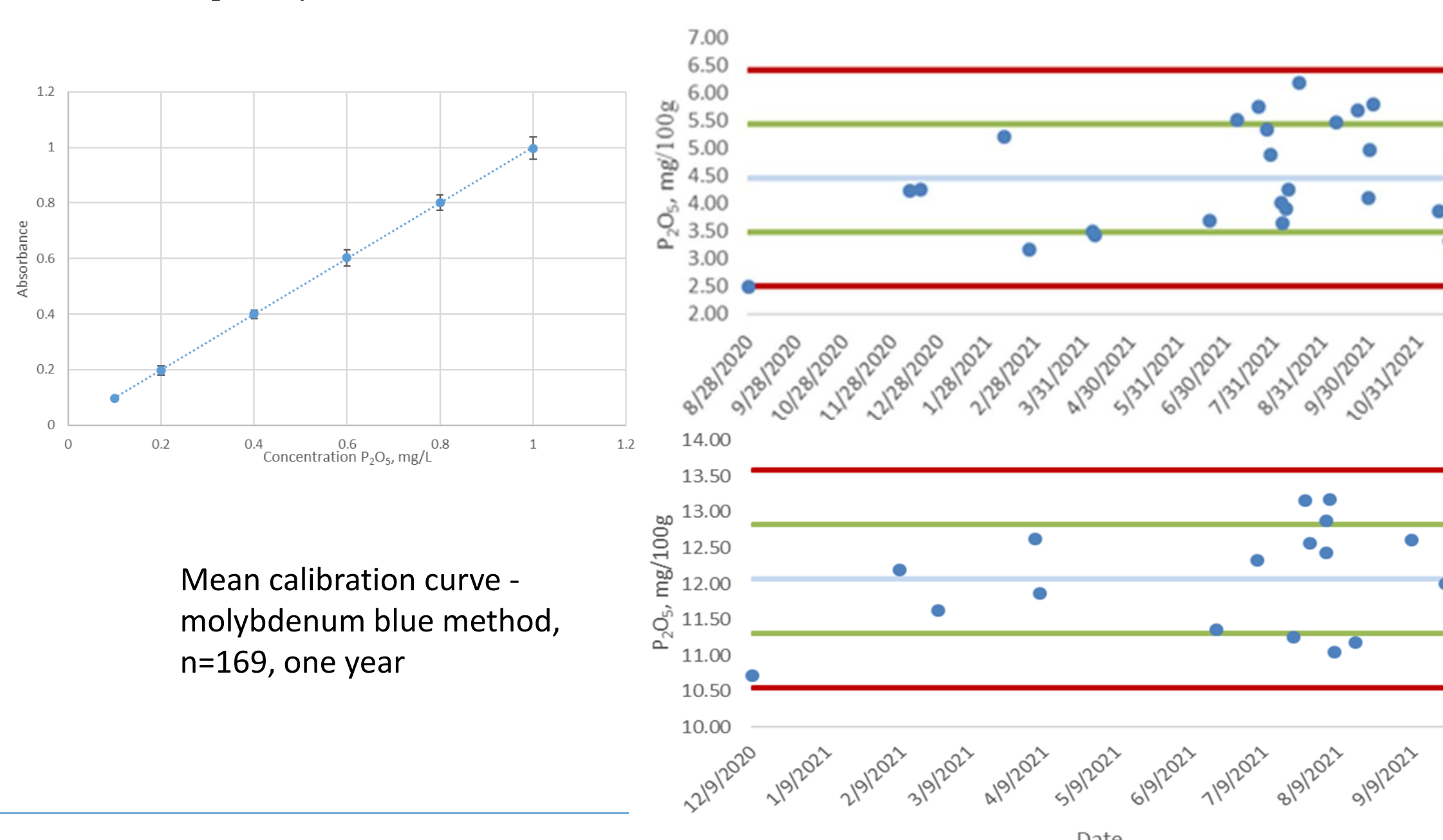
**Advantage:** addition before extraction - estimate bias of the whole procedure. The trueness to be evaluated in sample types usually encountered in the laboratory. The efficiency of extracting solution depends not only on its composition and procedure used, but on the soil type and its physical and chemical properties.

**Drawback:** the spiked analyte could not completely reach equilibrium with the soil sample, the obtained bias appeared approximate estimate of bias of available potassium in soil

**Solution:** standard addition after the extraction - estimate partially the method bias.

## Case 4: Quality control - in-lab control sample

To ensure the quality control, CRM and in-lab control samples were used in our laboratory. However, we found that the differences in particle size in standard and testing samples highly influenced the obtained results and sometimes they appeared non-informative to be applied for quality control. In the research laboratory typically one analyses limited number different samples in different conditions applying modified procedures which limits the obtaining of reasonable quantity of data for construction of useful control charts.



Control charts of arable soil sample containing (a) 4.6 and (b) 12.1 mg/100 g P<sub>2</sub>O<sub>5</sub>. The precision was estimated in within lab reproducibility conditions by standard deviation of concentration values obtained during one-year period by different analysts with different reagents, but the same instruments. The warning limits (green lines) were set at 1\*σ and the action limits (red line) at 2\*σ, blue line presented the mean value

## Heavy metals distribution in geochemical phases BCR procedure

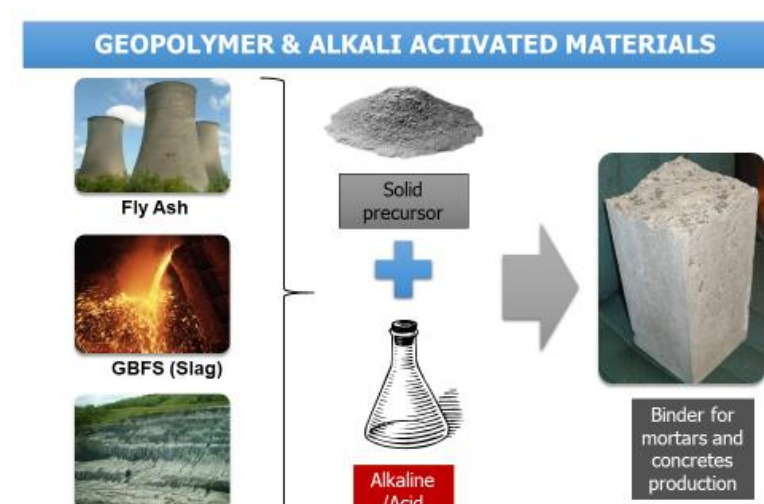
Fraction	Procedure
Fraction 1	• Mobile exchangeable fraction (exchangeable ions + carbonates) • 0.11 M CH <sub>3</sub> COOH, 16 h
Fraction 2	• Reducible fraction • 0.1 M NH <sub>2</sub> OH·HCl, pH 1.5 (HNO <sub>3</sub> ), 16 h
Fraction 3	• Oxidisable fraction • (a) 30 % H <sub>2</sub> O <sub>2</sub> , pH 2-3 (HNO <sub>3</sub> ), 2 h, t=85 °C • (b) 1 M CH <sub>3</sub> COONH <sub>4</sub> , pH 2 (HNO <sub>3</sub> ), 16 h
Fraction 4	• Residual fraction • Aqua regia digestion

## Heavy metals distribution in geochemical fraction of copper mine tailing

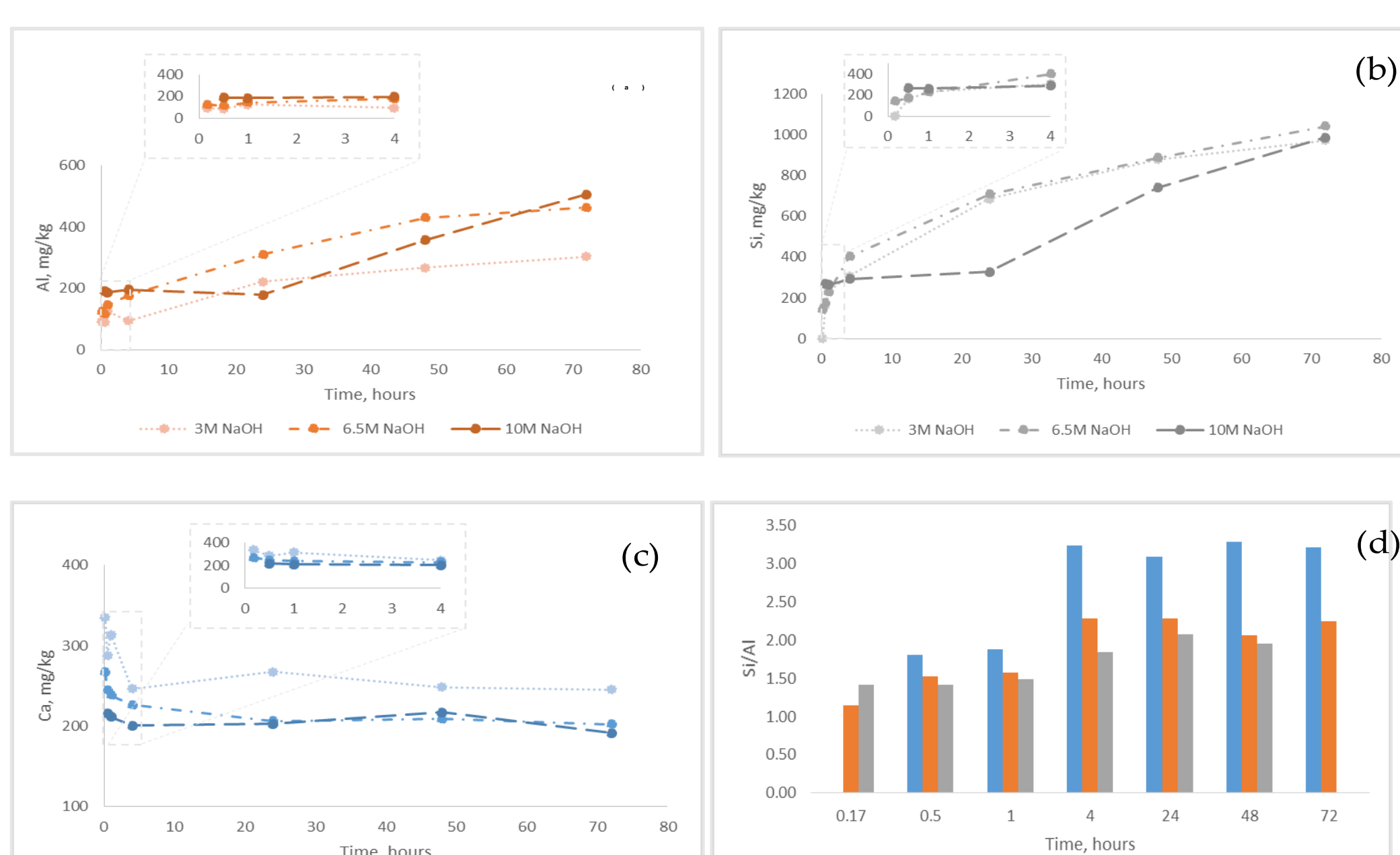
fraction	Fe	Cu	Zn	Mn	Cd	Ni	Pb
F1	447	107	5.5	20.5	< 0.25	< 0.5	3.8
F2	740	41.4	6.5	10.3	< 0.25	< 0.5	9.2
F3	8142	188	10.3	40.6	< 0.25	< 0.5	13.7
F1+F2+F3	9329	337	22.2	71.4	< 0.25	< 0.5	26.7
F4	10387	100	40.2	102	< 0.25	< 0.5	17.5
total	197116	437	62.4	173	< 0.25	< 0.5	44.2

Problems:  
Quality control  
BCR reference material - lake sediments  
Standard BCR procedure - needs to be scaled down

## Case 3: New specific procedure - alkaline reactivity of mine tailings and fly ash



Thirdly, when studying the behaviour of newly developed procedure for estimation of specific characteristics of the sample (for example, study of alkaline reactivity of mine tailing or fly ashes as precursors for geopolymer obtaining) the problems are even more pronounced. Applying the procedure to available CRM of soils or mine tailing doesn't provide the informative data as the reference or consensus values are not available for this specific procedure. An alternative approach was to study the behaviour of pure components and phases of the sample in the tested media. However, we found that we are scaling up the method and the obtained data could only roughly present the behaviour of the method in the tested materials.



## Alkaline reactivity

Problems:  
ICP-OES measurement  
High salt content  
High concentration of target elements  
Quality control  
Lack of reference material  
Lack of standard procedure



Solutions:  
ICP-OES measurement  
Dilution → Increased uncertainty  
Observation mode  
Quality control  
Interlaboratory comparison

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